## Data collection

| Enraf-Nonius CAD-4 single- | $\theta_{\max }=70^{\circ}$ |
| :--- | :--- |
| $\quad$ crystal diffractometer | $h=0 \rightarrow 16$ |
| $\omega / \theta$ scans | $k=0 \rightarrow 16$ |
| Absorption correction: none | $l=0 \rightarrow 18$ |
| 2928 measured reflections | 1 standard reflection |
| 2928 independent reflections | frequency: 120 min |
| 2647 reflections with | intensity decay: $1 \%$ |

## Refinement

Refinement on $F$
$(\Delta / \sigma)_{\max }=0.03$
$R=0.046$
$w R=0.063$
$S=1.870$
2647 reflections
307 parameters
H atoms not refined
$w=4 F_{o}^{2} /\left[\sigma^{2}\left(F_{o}^{2}\right)\right.$
$\left.+\left(0.04 F_{o}^{2}\right)^{2}\right]$
Table 1. Selected geometric parameters $\left(\AA,^{\circ}\right)$

| $\mathrm{N} 1-\mathrm{Cl} A$ | 1.412 (4) | $\mathrm{O}^{\prime}{ }^{\prime}-\mathrm{C} 3^{\prime}$ | 1.240 (3) |
| :---: | :---: | :---: | :---: |
| $\mathrm{N} 2-\mathrm{Cl}^{\prime}$ | 1.345 (4) | $\mathrm{O} 4^{\prime}-\mathrm{C} 4^{\prime}$ | 1.197 (5) |
| $\mathrm{N} 2-\mathrm{C} 2 A$ | 1.453 (3) | $\mathrm{Cl}^{\prime}-\mathrm{Cl} A$ | 1.504 (4) |
| N3-C2 ${ }^{\prime}$ | 1.342 (3) | $\mathrm{C} 2^{\prime}-\mathrm{C} 2 \mathrm{~A}$ | 1.522 (4) |
| N3-C3A | 1.427 (3) | C3'-C3A | 1.498 (4) |
| $\mathrm{N} 4-\mathrm{C}^{\prime}$ | 1.334 (3) | $\mathrm{C} 4^{\prime}-\mathrm{C} 4 \mathrm{~A}$ | 1.515 (4) |
| N4-C4A | 1.450 (4) | $\mathrm{Cl} A-\mathrm{Cl} B$ | 1.334 (4) |
| $\mathrm{Ol}^{\prime}-\mathrm{Cl}^{\prime}$ | 1.234 (3) | $\mathrm{C} 3 A-\mathrm{C} 3 B$ | 1.331 (4) |
| $\mathrm{O} 2^{\prime}-\mathrm{C}^{\prime}$ | 1.227 (3) |  |  |
| $\mathrm{C} 1^{\prime}-\mathrm{N} 2-\mathrm{C} 2 \mathrm{~A}$ | 120.2 (2) | $\mathrm{N} 2-\mathrm{C} 2 \mathrm{~A}-\mathrm{Cl}^{\prime}$ | 112.4 (2) |
| $\mathrm{C} 2^{\prime}-\mathrm{N} 3-\mathrm{C} 3$ A | 118.9 (2) | $\mathrm{N} 2-\mathrm{C} 2 A-\mathrm{C} 2 B$ | 109.7 (2) |
| C 3 - $\mathrm{N} 4-\mathrm{C} 4 \mathrm{~A}$ | 122.1 (2) | $\mathrm{N} 3-\mathrm{C} 3 \mathrm{~A}-\mathrm{C} 3^{\prime}$ | 115.5 (2) |
| $\mathrm{N} 2-\mathrm{Cl}^{\prime}-\mathrm{Cl} A$ | 116.1 (2) | N3-C3A-C3B | 124.9 (2) |
| $\mathrm{Ol}^{\prime}-\mathrm{Cl}^{\prime}-\mathrm{Cl} A$ | 121.6 (3) | N4-C4A-C4 ${ }^{\prime}$ | 113.1 (3) |
| $\mathrm{O} 2^{\prime}-\mathrm{C} 2^{\prime}-\mathrm{C} 2 \mathrm{~A}$ | 119.2 (2) | N4- $\mathrm{C} 4 A-\mathrm{C} 4 B$ | 107.9 (2) |
| $\mathrm{N} 4-\mathrm{C} 3^{\prime}-\mathrm{C} 3 A$ | 116.5 (2) | $\mathrm{Cl} A-\mathrm{Cl} B-\mathrm{ClC}$ | 128.6 (3) |
| $\mathrm{O} 3^{\prime}-\mathrm{C}^{\prime}-\mathrm{C} 3 \mathrm{~A}$ | 121.8 (2) | $\mathrm{C} 3 A-\mathrm{C} 3$ - $\mathrm{C} 3 C$ | 132.2 (2) |
| $\mathrm{N} 1-\mathrm{Cl} A-\mathrm{Cl}^{\prime}$ | 117.3 (2) | $\mathrm{C} 1 B-\mathrm{Cl} C-\mathrm{Cl} D$ | 117.8 (3) |
| $\mathrm{N} 1-\mathrm{Cl} A-\mathrm{Cl} B$ | 124.0 (3) | $\mathrm{C} 3 B-\mathrm{C} 3 C-\mathrm{C} 3 \mathrm{D}$ | 125.2 (3) |
| $\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 1 \mathrm{~A}-\mathrm{Cl}^{\prime}$ | -49.2 (4) | $\mathrm{C} 4 \mathrm{~A}-\mathrm{N} 4-\mathrm{C} 3^{\prime}-\mathrm{C} 3 \mathrm{~A}$ | 179.4 (2) |
| $\mathrm{Cl}^{\prime}-\mathrm{N} 2-\mathrm{C} 2 A-\mathrm{C}^{\prime}$ | -60.8 (3) | $\mathrm{N} 2-\mathrm{Cl}^{\prime}-\mathrm{Cl} A-\mathrm{N} 1$ | -28.0 (3) |
| $\mathrm{C} 2^{\prime}-\mathrm{N} 3-\mathrm{C} 3 \mathrm{~A}-\mathrm{C} 3^{\prime}$ | -56.5 (3) | $\mathrm{Ol}^{\prime}-\mathrm{Cl}^{\prime}-\mathrm{Cl} A-\mathrm{ClB}$ | -34.2(4) |
| $\mathrm{C} 3^{\prime}-\mathrm{N} 4-\mathrm{C} 4 \mathrm{~A}-\mathrm{C} 4^{\prime}$ | 48.0 (3) | $\mathrm{N} 3-\mathrm{C} 2^{\prime}-\mathrm{C} 2 \mathrm{~A}-\mathrm{N} 2$ | -24.1 (4) |
| $\mathrm{C} 2 \mathrm{~A}-\mathrm{N} 2-\mathrm{Cl}{ }^{\prime}-\mathrm{C} 1 A$ | 179.5 (2) | $\mathrm{N} 4-\mathrm{C} 3^{\prime}-\mathrm{C} 3 \mathrm{~A}-\mathrm{N} 3$ | -30.6 (3) |
| $\mathrm{C} 3 \mathrm{~A}-\mathrm{N} 3-\mathrm{C}^{\prime}-\mathrm{C} 2 A$ | 173.9 (2) | $\mathrm{O} 2-\mathrm{C} 4{ }^{\prime}-\mathrm{C} 4 A-\mathrm{N} 4$ | 40.0 (4) |

Table 2. Hydrogen-bonding geometry $\left(\AA,{ }^{\circ}\right)$

| D—H. ${ }^{\text {ch }}$ | $D-\mathrm{H}$ | H...A | D.. A | D-H. . A |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N} 1-\mathrm{HN} 1 \cdots \mathrm{O}^{\prime \prime}{ }^{\text {i }}$ | 0.96 | 1.950 (2) | 2.867 (3) | 160.1 (1) |
| $\mathrm{N} 2-\mathrm{HN} 2 \cdots \mathrm{O}^{\text {/i }}$ | 0.92 | 1.986 (2) | 2.856 (3) | 156.7 (1) |
| N3-HN3..Ol | 0.86 | 2.087 (2) | 2.932 (3) | 169.1 (2) |
| N4-HN4...O1' | 0.97 | 1.976 (2) | 2.949 (3) | 175.7 (1) |

H atoms attached to C atoms were placed at idealized positions, while those bonded to N atoms were located from difference Fourier maps. During refinement, all H atoms were allowed to ride with isotropic displacement parameters set at $U_{\text {eq }}$ of the carrier atom.
Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: SDP (Enraf-Nonius, 1985). Data reduction: $S D P$. Program(s) used to solve structure: SIR92 (Altomare et al., 1993) (direct methods). Program(s) used to refine structure: SDP. Molecular graphics: ORTEPII (Johnson, 1976) and PLUTO (Motherwell \& Clegg, 1978).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: SX1026). Services for accessing these data are described at the back of the journal.

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# Diastereoselectivity in the Transannular Diels-Alder Reaction of a trans-transtrans 14-Membered Macrocycle Leading to Steroids 

Marc Drouin, ${ }^{a}$ Michel Couturier ${ }^{b}$ and Pierre Deslongchamps ${ }^{b}$
${ }^{a}$ Laboratoire de diffraction des rayons- $X$, Département de chimie, Faculté des Sciences, Université de Sherbrooke, Sherbrooke, Québec, Canada JIK 2RI, and ${ }^{b}$ Laboratoire de synthèse organique, Département de chimie, Faculté des Sciences, Université de Sherbrooke, Sherbrooke, Québec, Canada J1K2R1.E-mail: mdrouin@courrier.usherb.ca
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## Abstract

A novel 14 -membered macrocyclic ring with trans-trans-trans triene geometry, trimethyl ( $4 E, 10 E, 12 E$ )( $1 R^{*}, 14 S^{*}$ )-5-methyl-17-oxobicyclo[12.3.0]heptadeca-

4,10,12-triene-1,8,8-tricarboxylate, $\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{O}_{7}$, which usually undergoes a Diels-Alder cycloaddition upon formation, has been isolated. From four possible contractions, the transannular reaction produces three adducts from which trimethyl rac-( $5 \beta, 9 \beta, 10 \alpha$ )-17-oxoandrost6 -ene-3,3,18-tricarboxylate, $\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{O}_{7}$, was isolated and crystallized.

## Comment

In the course of our general study on the transannular Diels-Alder (TADA) reaction involving 14 -membered macrocyclic trienes (Deslongchamps, 1992), we demonstrated that the trans-trans-trans (TTT) geometry leads to the cis-anti-trans (CAT) and trans-anti-cis (TAC) [6.6.6] adducts, the former being comparable to the $A B C 5 \beta$-steroids stereochemistry (Ndibwami, Lamothe, Soucy, Goldstein \& Deslongchamps, 1993). Application of the above strategy to steroid total synthesis was undertaken with a macrocycle containing a trans-fused $D$ ring. In the present case, however, such a macrocycle, (1), can potentially collapse to four different adducts, namely trans-anti-cis-anti-trans (TACAT), (2), cis-anti-trans-syn-trans (CATST), (3), trans-anti-cis-syn-trans (TACST), (4), and cis-anti-trans-anti-trans (CATAT), (5).

(1')

(1)

$$
E=\mathrm{CO}_{2} \mathrm{CH}_{3}
$$


(2)

(4)

(3)

(5)

Since the ratio of final products in the TADA reaction is governed by the energy gap between the corresponding transition-state structures with the incipient
$B$ ring in a boat conformation (Lamothe, Ndibwami \& Deslongchamps, 1988a,b), we used AM1 semi-empirical molecular modelling to predict the stereochemical outcome. Forecasting the CATAT stereoisomer, (5), as the major adduct, the in situ generation of macrocycle (1) from allylic chloride ( $1^{\prime}$ ) was performed which directly lead to a mixture of three adducts in a $4: 1: 1$ ratio. The major isomer was isolated by crystallization and an X-ray diffraction analysis was undertaken in order to establish unequivocally its relative stereochemistry: it turned out to be the TACAT stereoisomer, (2). This unexpected result prompted us to investigate further with more reliable $a b$ initio transition-state modelling and these calculations now corroborate the observed diastereoselectivity (Couturier, Dory, Rouillard, Fortin \& Deslongchamps, 1997). During the macrocyclization process in the TTT 14-membered series, the macrocycles are generally not isolated since the ensuing TADA reaction occurs at the reaction temperature of 353 K . However, by deliberately halting the reaction before completion, we were able to isolate, albeit in minute quantities, the intermediate carbocycle (1), which was crystallized, and we were at last able to determine the macrocycle's conformation in this series.

In compound (1), the methyl ester located at the ring junction shows disorder. Successive $\Delta F$ maps result in three different orientations. The occupancy refinement converged at values of 0.48 (2) for C23, C24, O5 and O6, 0.23 (2) for C23A, C24A, O5A and O6A, and 0.29 (2) for $\mathrm{C} 23 B, \mathrm{C} 24 B$, O5B and $\mathrm{O} 6 B$. The torsion angle values for $\mathrm{C} 14-\mathrm{C} 13-\mathrm{C} 23-\mathrm{O}, \mathrm{C} 14-\mathrm{C} 13-$ $\mathrm{C} 23 A-\mathrm{O} 5 A$ and $\mathrm{C} 14-\mathrm{C} 13-\mathrm{C} 23 B-\mathrm{O} B B$ indicate the differences in their orientations [ -62.5 (15), 11 (2) and $-31(2)^{\circ}$, respectively]. The ester is oriented such that O5 or O5B could act as an acceptor for C$\mathrm{H} \cdots \mathrm{O}$ intramolecular hydrogen bonding. The $\mathrm{O} \cdots \mathrm{H}$ distances and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ angle values are $2.24 \AA$ and $122.8^{\circ}$ for $\mathrm{C} 11-\mathrm{H} 11 A \cdots \mathrm{O}$, and $2.29 \AA$ and $112.7^{\circ}$ for $\mathrm{C} 8-\mathrm{H} 8 \cdots \mathrm{O} 5$, the two most populated conformations. These distances and angles are in good agreement


Fig. 1. ORTEP (Johnson, 1995) perspective view showing the labelling for compound (1). Displacement ellipsoids are shown at the $30 \%$ probability level; H atoms have been omitted for clarity except for those on $s p^{2}-\mathrm{C}$ atoms and those on ring junctions which are drawn as small circles of arbitrary radii. Only one orientation for the disordered methyl ester groups is retained.
with those described by Desiraju, Kashino, Coombs \& Glusker (1993). The diene shows a partially broken conjugated system. Indeed, the torsion angle value for C5-C6-C7-C8 is $-160.9(3)^{\circ}$, which is a good indication of the strain in the macrocycle.

Compound (2) crystallizes with two molecules per asymmetric unit. Both skeletons have the same configuration as well as similar global conformations. This chiral crystal structure arises from spontaneous resolution of a racemic mixture. Its absolute configuration has not been determined. One of the two molecules, ( $2^{\prime}$ ), shows disorder in rings $A$ and $D$. The occupancy refinement. converged at $0.553(7): 0.447(7)$ for $\mathrm{Cl}^{\prime}-\mathrm{C}^{\prime}$ and $\mathrm{ClB}-\mathrm{C} 4 B$, and $0.58(3): 0.42(3)$ for $\mathrm{C} 15^{\prime}, \mathrm{C} 16^{\prime}$ and $\mathrm{C} 15 B, \mathrm{C} 16 B$. The most important deviations are for the five-membered ring $D$. Indeed, the torsion angles $\mathrm{C} 15-$ $\mathrm{C} 16-\mathrm{Cl} 7-\mathrm{C} 13$ and $\mathrm{C} 12-\mathrm{C} 13-\mathrm{Cl}-\mathrm{C} 16$ have values $6.8(4)$ and $-149.6(4)^{\circ}$, respectively, in (2), $\mathrm{Cl}^{\prime}$ $\mathrm{C}^{\prime} 6^{\prime}-\mathrm{C}_{1} 7^{\prime}-\mathrm{Cl}^{\prime}$ and $\mathrm{C} 12^{\prime}-\mathrm{C}_{1}^{\prime}-\mathrm{C}^{\prime} 7^{\prime}-\mathrm{C}_{1} 6^{\prime}$ are $-16.2(10)$ and $-135.9(7)^{\circ}$, respectively, in ( $2^{\prime}$ ), and $\mathrm{C} 15 B-\mathrm{C} 16 B-\mathrm{Cl7}^{\prime}-\mathrm{Cl3}^{\prime}$ and $\mathrm{Cl2}^{\prime}-\mathrm{Cl3}^{\prime}-\mathrm{Cl}^{\prime}-$ $C 16 B$ are 36.3 (15) and -157.9 (7), respectively, in $\left(2^{\prime} B\right)$. The three methyl esters in this molecule are also disordered. In the crystal packing, the molecules form two different layers. The disorder is apparently caused by an intermolecular close contact between C22' and


Molecule (2)


Molecule (2')
Fig. 2. ORTEP (Johnson, 1995) perspective view showing the labelling for molecules (2) and ( $2^{\prime}$ ). Displacement ellipsoids are shown at the $30 \%$ probability level; H atoms have been omitted for clarity except for those on $s p^{2}-\mathrm{C}$ atoms and those on ring junctions which are drawn as small circles of arbitrary radii. Only one orientation for the disordered methyl ester groups is retained for ( $2^{\prime}$ ).

C20B [2.580 (16) Å]. To avoid this contact, molecule ( $2^{\prime}$ ) adopts the two conformations observed. All bonds and angles have normal values.

## Experimental

Details of the synthesis will be published elsewhere (Couturier, 1997; Couturier, Dory, Rouillard, Fortin \& Deslongchamps, 1997).

## Compound (1)

Crystal data
$\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{O}_{7}$
$M_{r}=432.51$
Monoclinic
A2/a
$a=23.756(2) \AA$
$b=6.9151$ (5) $\AA$
$c=28.381$ (2) $\AA$
$\beta=95.457(6)^{\circ}$
$V=4641.1(6) \AA^{3}$
$Z=8$
$D_{x}=1.238 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ not measured

## Data collection

Nonius CAD-4 diffractom-
eter
$\theta / 2 \theta$ scans
Absorption correction: none
4819 measured reflections
4279 independent reflections 2600 reflections with
$I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.060$
$\omega R\left(F^{2}\right)=0.176$
$S=1.071$
4279 reflections
356 parameters
H atoms riding
$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0797 P)^{2}\right.$
$+0.0298 P$ ]
where $P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3$
$\mathrm{Cu} K \alpha$ radiation
$\lambda=1.54184 \AA$
Cell parameters from 24
reflections
$\theta=20-25^{\circ}$
$\mu=0.742 \mathrm{~mm}^{-1}$
$T=293 \mathrm{~K}$
Rectangular
$0.20 \times 0.15 \times 0.05 \mathrm{~mm}$
Colourless
$R_{\text {int }}=0.023$
$\theta_{\text {max }}=69.85^{\circ}$
$h=-28 \rightarrow 28$
$k=0 \rightarrow 7$
$l=0 \rightarrow 34$
2 standard reflections every 60 reflections intensity decay: none
$(\Delta / \sigma)_{\text {max }}=-0.001$
$\Delta \rho_{\text {max }}=0.267 \mathrm{e}^{-3}$
$\Delta \rho_{\text {min }}=-0.243 \mathrm{e} \mathrm{A}^{-3}$
Extinction correction: SHELXL93
Extinction coefficient: 0.00016 (6)

Scattering factors from International Tables for Crystallography (Vol. C)

Compound (2)
Crystal data
$\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{O}_{7}$
$M_{r}=432.50$
Monoclinic
$P 21$
$a=7.278$ (4) $\AA$
$b=12.199$ (2) $\AA$
$c=24.494$ (7) $\AA$
$\beta=91.89(3)^{\circ}$
$V=2173.6(13) \AA^{3}$
$Z=4$
$D_{x}=1.322 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ not measured

## $\mathrm{Cu} K \alpha$ radiation

$\lambda=1.54184 \AA$
Cell parameters from 24 reflections
$\theta=20-25^{\circ}$
$\mu=0.793 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Colourless
$0.35 \times 0.20 \times 0.10 \mathrm{~mm}$
Rectangular

## Data collection

Nonius CAD-4 diffractometer
$\theta / 2 \theta$ scans
Absorption correction: none
4211 measured reflections
4211 independent reflections
4010 reflections with
$I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.060$
$w R\left(F^{2}\right)=0.179$
$S=1.062$
4211 reflections
665 parameters
H atoms riding

$$
\begin{aligned}
& w= 1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.1137 P)^{2}\right. \\
&+1.7798 P] \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=-0.013
\end{aligned}
$$

$$
\begin{aligned}
& \theta_{\max }=71.74^{\circ} \\
& h=-6 \rightarrow 8 \\
& k=0 \rightarrow 14 \\
& l=0 \rightarrow 30 \\
& 2 \text { standard reflections } \\
& \text { frequency: } 60 \text { min } \\
& \text { intensity decay: none }
\end{aligned}
$$

$$
\Delta \rho_{\max }=0.323 \mathrm{e}_{\circ}^{-3}
$$

$$
\Delta \rho_{\min }=-0.250 \text { e } \AA^{-3}
$$

Extinction correction:
SHELXL93

Extinction coefficient: 0.0014 (4)

Scattering factors from International Tables for Crystallography (Vol. C)
Absolute configuration: Flack (1983)
Flack parameter $=-0.2(3)$

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# Dimethyl $N, N^{\prime}$-Bis(endo-himmoyl)-( $\boldsymbol{R}, \boldsymbol{R}$ )cystine $\dagger$ 

David E. Hibbs, ${ }^{a}$ Michael B. Hursthouse, ${ }^{a}$ K. M. Abdul Malik ${ }^{a}$ and Michael North ${ }^{b}$<br>${ }^{a}$ Department of Chemistry, University of Wales Cardiff, PO Box 912, Park Place, Cardiff CF1 3TB, Wales, and ${ }^{b}$ Department of Chemistry, University of Wales Bangor, Gwynedd LL57 2UW, Wales. E-mail: sackam@cardiff.ac.uk

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The disordered atoms in (1) were refined using the SAME, SADI and SIMU restraints in SHELXL93 (Sheldrick, 1993); the disorded atoms in (2') were refined using SAME and SADI restraints. The $\mathrm{C} 15^{\prime}, \mathrm{C} 15 B, \mathrm{C} 16^{\prime}, \mathrm{C} 16 B$ and methyl ester disordered atoms at C13 were kept isotropic to avoid unrealistic displacement parameters.

For both compounds, data collection: NRCCAD DATCOL (Le Page, White \& Gabe, 1986); cell refinement: NRCCAD TRUANG; data reduction: NRCVAX DATRD2 (Gabe, Le Page, Charland, Lee \& White, 1989); program(s) used to solve structures: NRCVAX SOLVER; program(s) used to refine structures: SHELXL93 (Sheldrick, 1993); molecular graphics: ORTEP in Xtal_GX (Johnson, 1995); software used to prepare material for publication: SHELXL93 ACTA.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1287). Services for accessing these data are described at the back of the journal.

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Le Page, Y., White, P. S. \& Gabe, E. J. (1986). NRCCAD. An Enhanced CAD-4 Control Program. Proc. Am. Crystallogr. Hamilton Meet. Abstr. PA23.

## Abstract

The title compound, $\mathrm{C}_{26} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{8} \mathrm{~S}_{2}$, contains two norbornene rings, both with endo substituents, and an ( $M$ )-helical disulfide. Both ester groups adopt the $s$-cis conformation, and the bond lengths and angles are within the expected values.

## Comment

As part of an ongoing study concerned with the synthesis of biomimetic polymers derived from amino acids (Coles et al., 1994; Biagini et al., 1995), the synthesis of the title compound, (I), was undertaken (Biagini et al., 1995). It was envisaged that polymerization or copolymerization of (I) by a ring-opening metathesis procedure would lead to synthetic polymers in which the cystine units mimic the crosslinking role of cystine residues in proteins. An X-ray structure determination of (I) was undertaken to allow its conformation to be compared with those of other cystine derivatives.


This study confirmed the structure (Fig. 1) and relative stereochemistry of (I). The imide substituents on both norbornene rings adopt the endo configuration, and the two chiral centres ( C 10 and $\mathrm{C} 10^{\prime}$ ) have the same

[^0]
[^0]:    $\dagger$ Alternative systematic name: dimethyl 2,7-bis(1,3-dioxo-1,3,3a,4,7,7a-hexahydro-4,7-methano-2-isoindolyl)-4,5-dithiaoctanedioate.

